

NOx Removal Processes by Microplasma Generation in Multiple Electrode Configuration



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Introduction

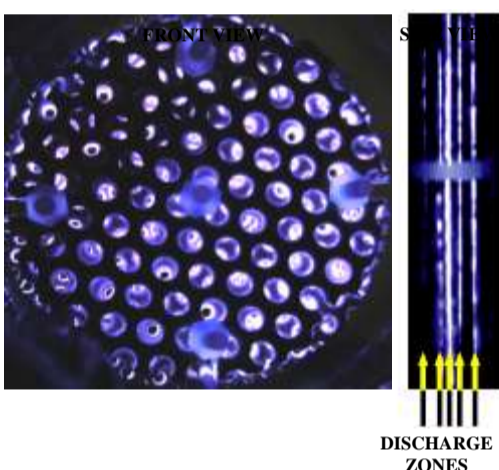
Nonthermal plasma was investigated in the decade as a solution to the environmental problems generated by the exhaust gases from factories, automobiles and other sources. High energy electrons and active species generated by non thermal plasma, could transform pollutants from exhaust gases in simpler compounds.

Microplasma is a nonthermal plasma at atmospheric pressure generated at very small discharge gaps and low discharge voltages. NOx removal from simulated exhaust gas is investigated by using microplasma.

Experimental Setup

(1) Microplasma Electrodes

The electrodes consist in perforated metallic plates covered with a dielectric layer.



Microplasma electrodes in multiple electrodes configuration

6 electrodes with \varnothing 42 mm faced together with a discharge gap of $0 \mu\text{m}$ \Rightarrow 5 discharge zones.

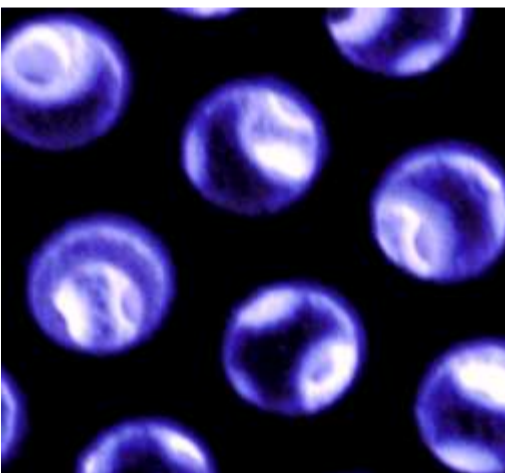
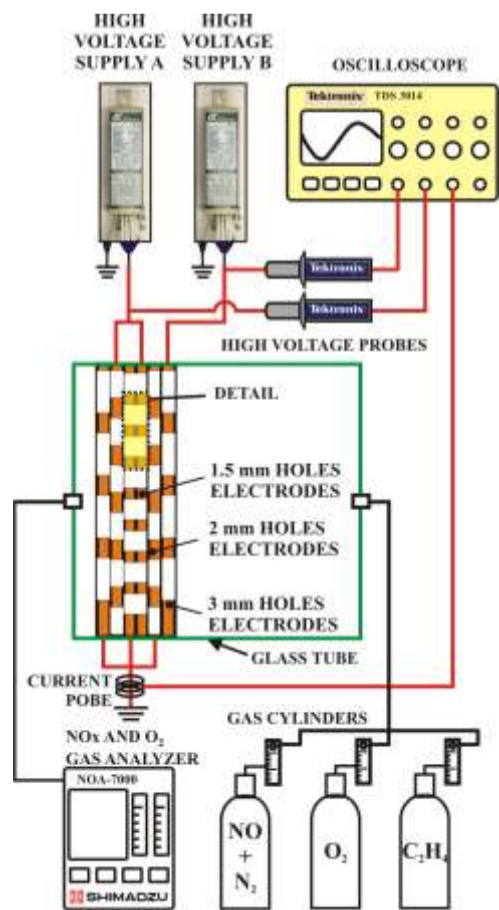


Photo of the microplasma discharge

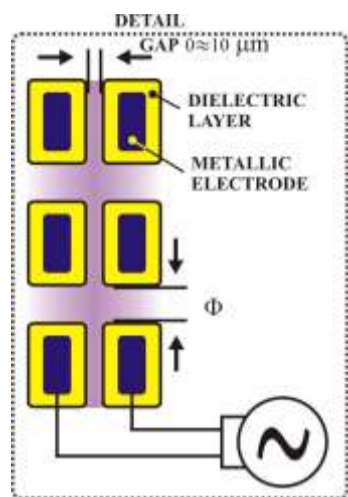
(2) Experimental Setup

Three discharge zones are supplied with voltage from one power supply: discharge voltage A.

Two discharge zones are supplied from the second power supply: discharge voltage B.



Schematic view of microplasma electrodes and the supply circuit



Detail of microplasma electrodes

Two small size power supplies (LECIP, M-1H) to supply to the electrodes alternative voltages with frequencies around 25 kHz.

Discharge gas: NO 500 ppm, C₂H₄ 3500 ppm, O₂ 5% and N₂ balance.

Gas flow rate: 5 L/min.

Aperture of electrodes:

- Middle pair 1.5 mm holes
 - Intermediate pair 2 mm holes
 - Extremities 3 mm holes
- \Rightarrow honeycomb like structure

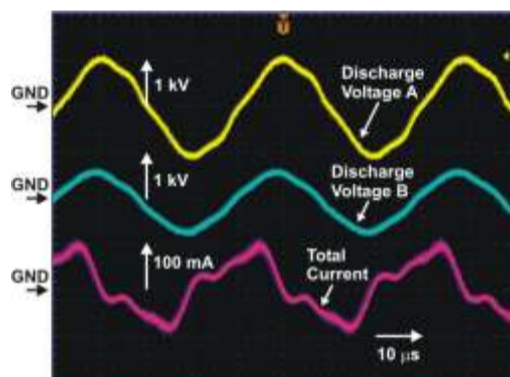
The treated gas was analyzed by a NOx and O₂ gas analyzer (Shimadzu NOA-7000).

NOx Removal Processes

(1) Electrical Characteristics

Very small discharge gaps and relatively low discharge voltages (about 1 kV)

\Rightarrow A high intensity electric field (10^7 - 10^8 V/m) assures the formation of microplasma and a corresponding discharge current.



Waveforms of the discharge voltages and discharge current

(2) NOx Removal

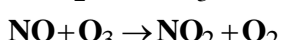
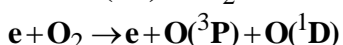
The highest values were obtained at 1.2 kV discharge voltage A and 1.2 kV discharge voltage B.

- NOx removal 51 ppm
- NO removal 218 ppm
- Generated NO₂ 171 ppm

NOx removal concentration, NO removal concentration and generated NO₂ concentration for different values of the discharge voltages

Discharge Voltage A (kV)	Discharge Voltage B (kV)	NO Removal Concentration (ppm)	NOx Removal Concentration (ppm)	NO ₂ Concentration (ppm)
0.7	0.7	45	9	42
0.7	1.2	113	21	102
0.7	0.8	53	17	47
0.7	0.9	61	19	52
0.7	1	79	21	73
0.7	1.1	94	26	81
0.9	0.9	120	34	101
0.9	0.9	135	28	111
0.9	0.9	133	35	110
0.9	1.2	155	35	136
0.95	0.95	160	39	126
0.95	0.95	167	47	140
0.95	0.95	169	44	132
1.1	0.7	147	43	124
1.1	1.1	171	51	136
1.1	1.2	195	48	158
1.2	0.7	181	48	141
1.2	1.2	218	51	171

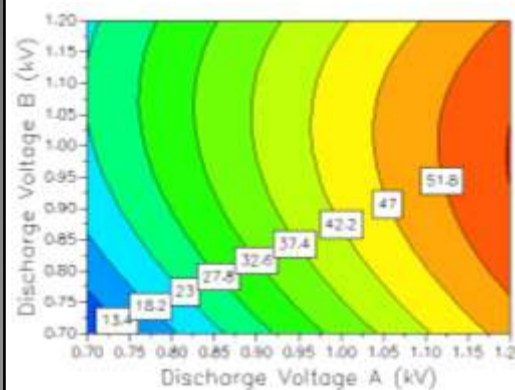
The dissociation energy for N₂ is 9 eV and for O₂ is 5-6 eV \Rightarrow with the increase of discharge voltage the NOx removal concentration will be higher:



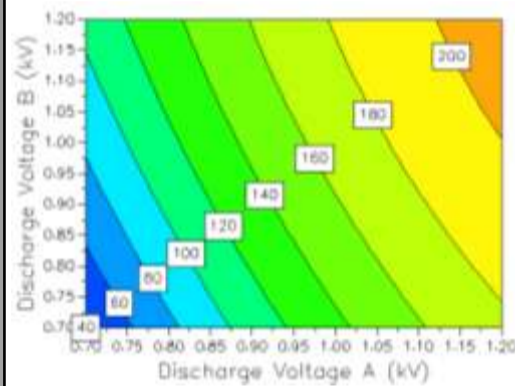
(3) Design of Experiments

The methodology of design of experiments proposes a factorial experimentation in which the factors

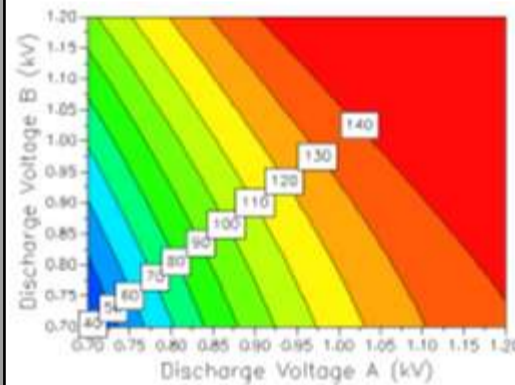
- Discharge voltage A
 - Discharge voltage B
- vary simultaneously.



Contour plot of the predicted response of NOx removal concentration



Contour plot of the predicted response of NO removal concentration



Contour plot of the predicted response of generated NO₂ concentration

Using MODDE 5.0 software the data were analyzed.

Larger microplasma area powered by high voltage power supply A:

\Rightarrow optimal predicted value of NOx removal concentration: 1.2 kV for discharge voltage A and 1 kV for discharge voltage B.

\Rightarrow optimal predicted value of NO removal concentration: 1.2 kV for discharge voltage A and 1.2 kV for discharge voltage B.

\Rightarrow optimal predicted value of generated NO₂ concentration: 1.2 kV for discharge voltage A and 0.95 kV to 1.2 kV for discharge voltage B.

Conclusions

•As the discharge voltages were increased the NOx removal concentration, NO removal concentration and generated NO₂ concentration increased.

•Design of experiments software analysis shows that with the increase of both discharge voltages higher values of the NOx and NO removal concentrations and generated NO₂ concentration are obtained.